## Bench Trials Acid Adjustment:

Here's a guide on running a bench trial to determine an acid adjustment for a high pH wine:

1. Mix a $10 \%$ Tartaric Acid solution by diluting 10 grams of Tartaric acid into 100 ml 's of DI water.
2. Fill 5 wine glasses with 100 ml 's of the wine in question.
3. Measure the wines pH and recording your findings.
4. Using a graduated pipette, drip $1 / 2 \mathrm{ml}$ of the TA solution in the first glass, 1 ml in the second, 1.5 ml in the third, and 2 ml in the fourth wine glass. Stir the solution into the wine and measure and record the pH of each glass.

Next taste each glass and determine which addition you prefer. The ml's of solution equals the grams per liter of tartaric acid to be added to your barrel, carboy, or bulk wine. Use the non-adjusted $5^{\text {th }}$ glass as a baseline for a taste comparison.

Example: If you prefer the third glass with the 1.5 ml addition then 1.5 grams per liter provides the equivalent result to your barrel (ex. 225 liter barrel would require $1.5 \times 225=337$ grams of Tartaric Acid). Depending on how high the pH is, it may take several rounds of additions before the wine stabilizes at your desired result. Be patient and test frequently.

| Mix up a 10\% TA Solution (10 grams Tartaric Acid into 100ml's of Distilled Water). <br> Example: Wine pH 4.01 to high. Susceptible to spoilage and <br> likely lacking varietal character. Reducing the pH will improve |  |  |  |  |
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| flavor and stability. |  |  |  |  |

Addition Equation: Number of ml's of TA solution = grams per liter to add to your wine.
Example:

- 225 liter barrel ( 59 gallon) to be adjusted.
- Selected glass with 1.5 ml TA addition with and example pH of 3.65 .
- $1.5 \mathrm{ml} \times 225$ liters $=337.5$ grams to add to barrel

Acid has a tendency to fall out of high pH wines so repeat the bench trial until you've attained the desired results.

Cheers!
The Vine, The Time,
The Wine。

